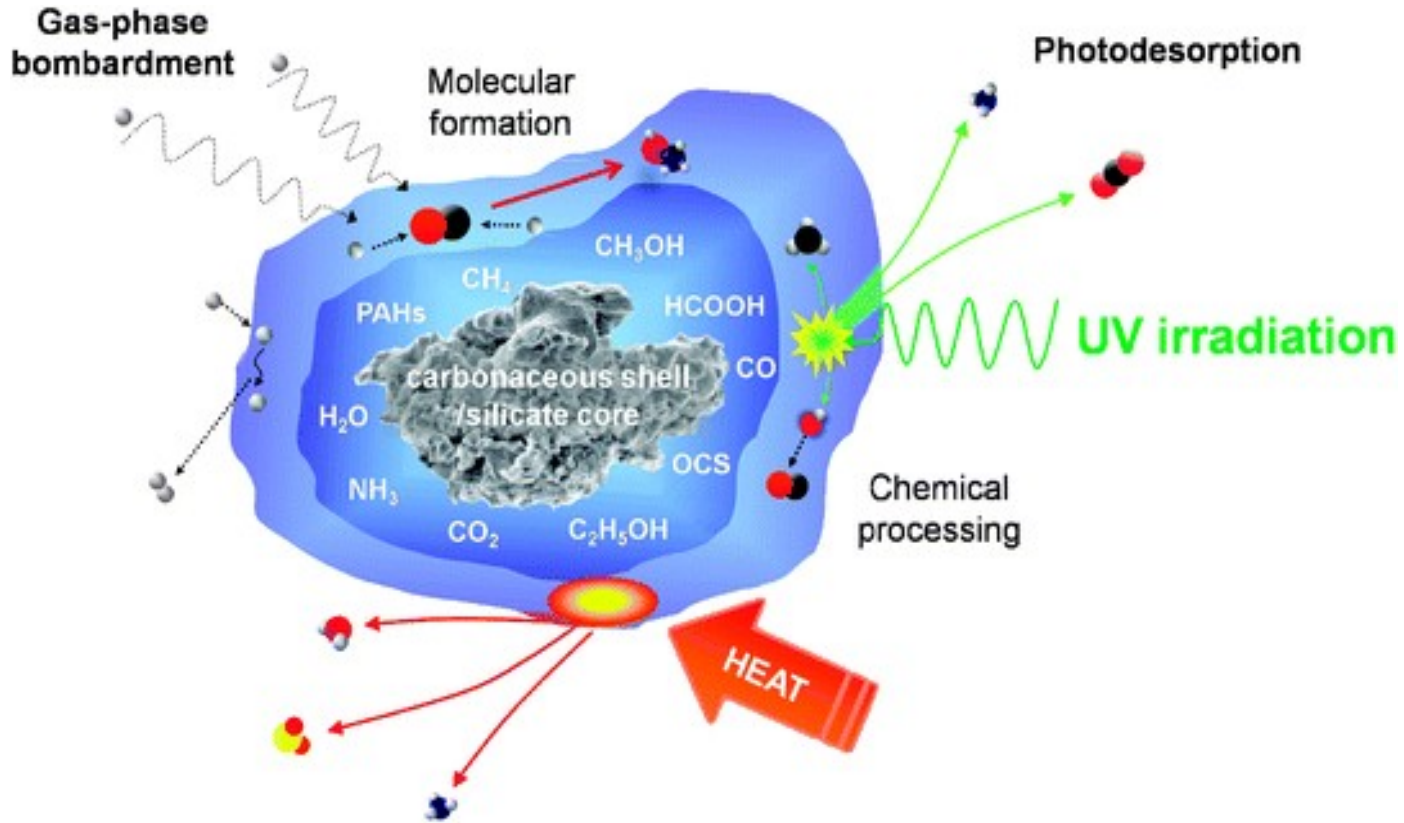


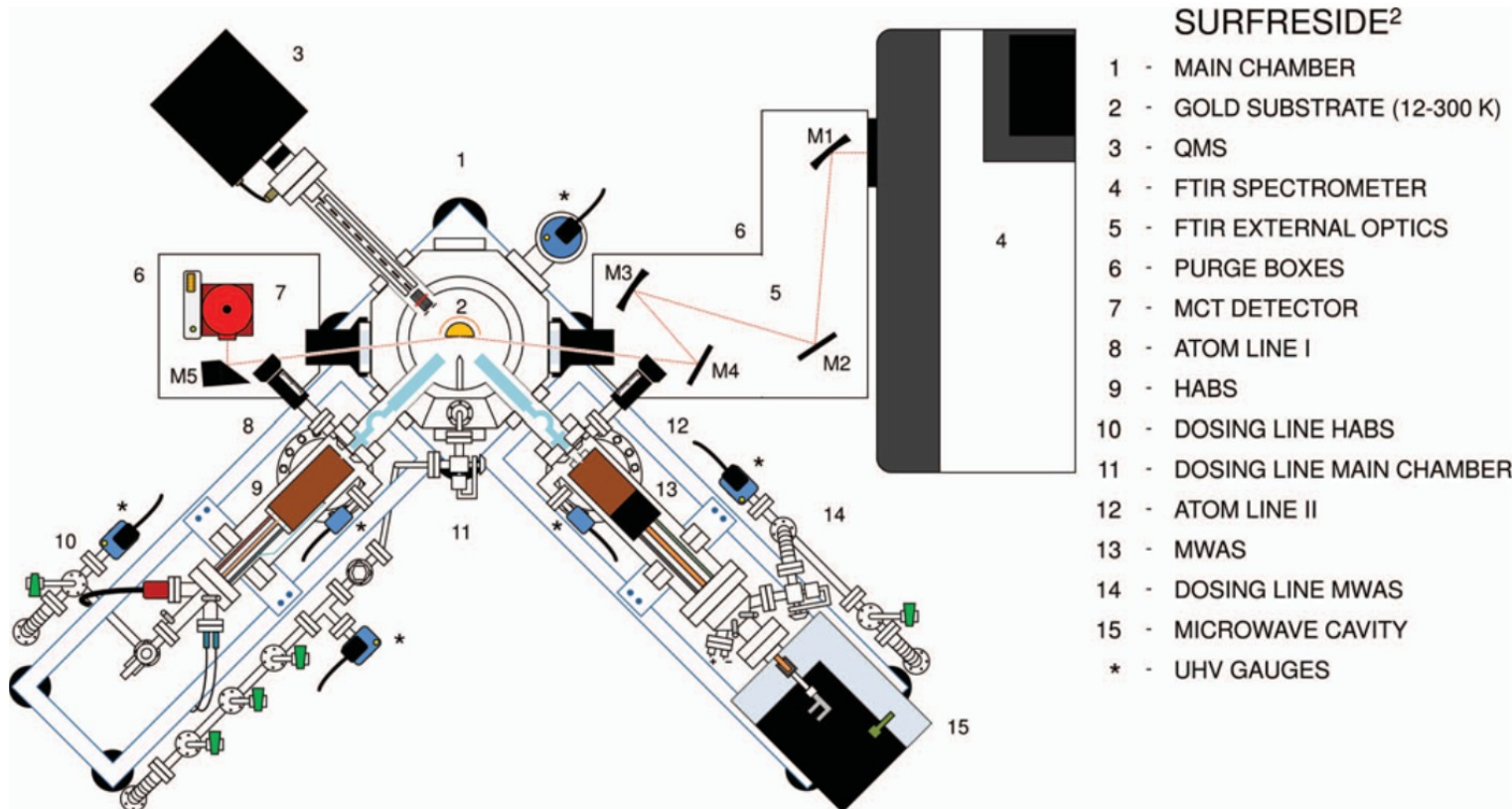
# Molecular complexity in interstellar ices

Jiao He  
Sackler Lab for Astrophysics  
Leiden Observatory  
November 28<sup>th</sup>, 2018  
DAN-II meeting at Leiden

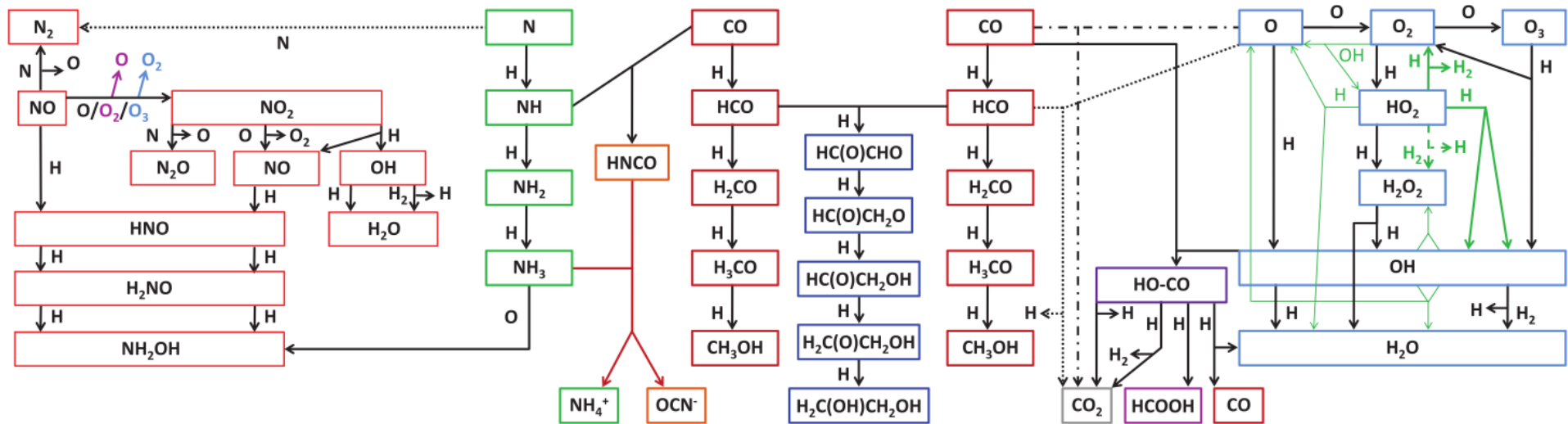
# Chemistry in the ice mantle



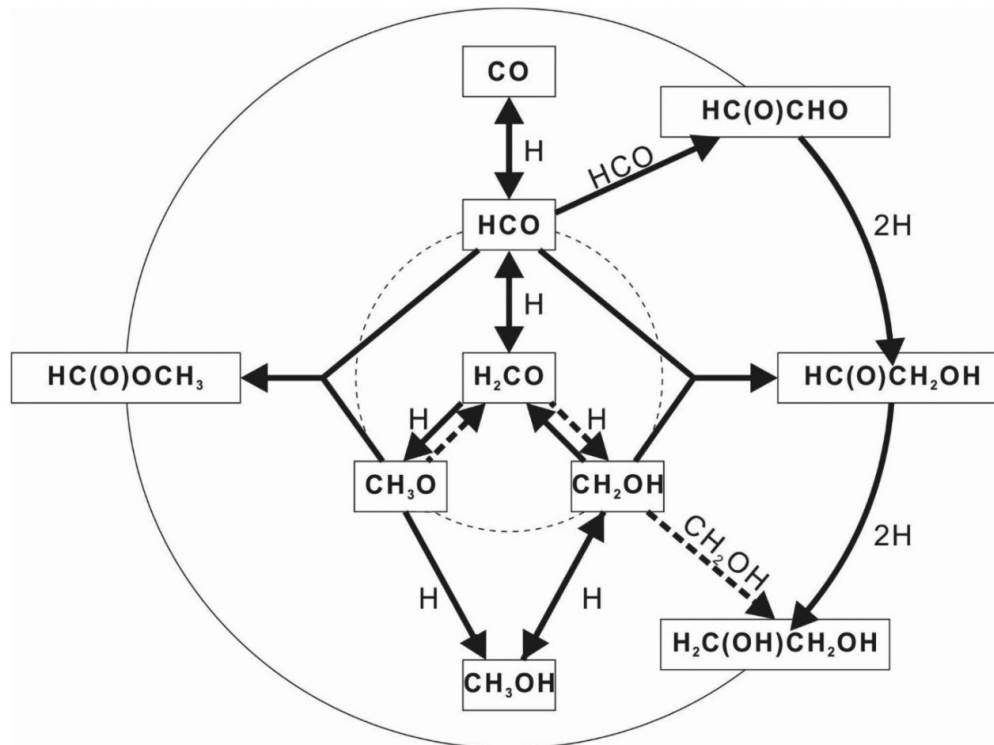
# The SURFRESIDE<sup>2</sup> apparatus



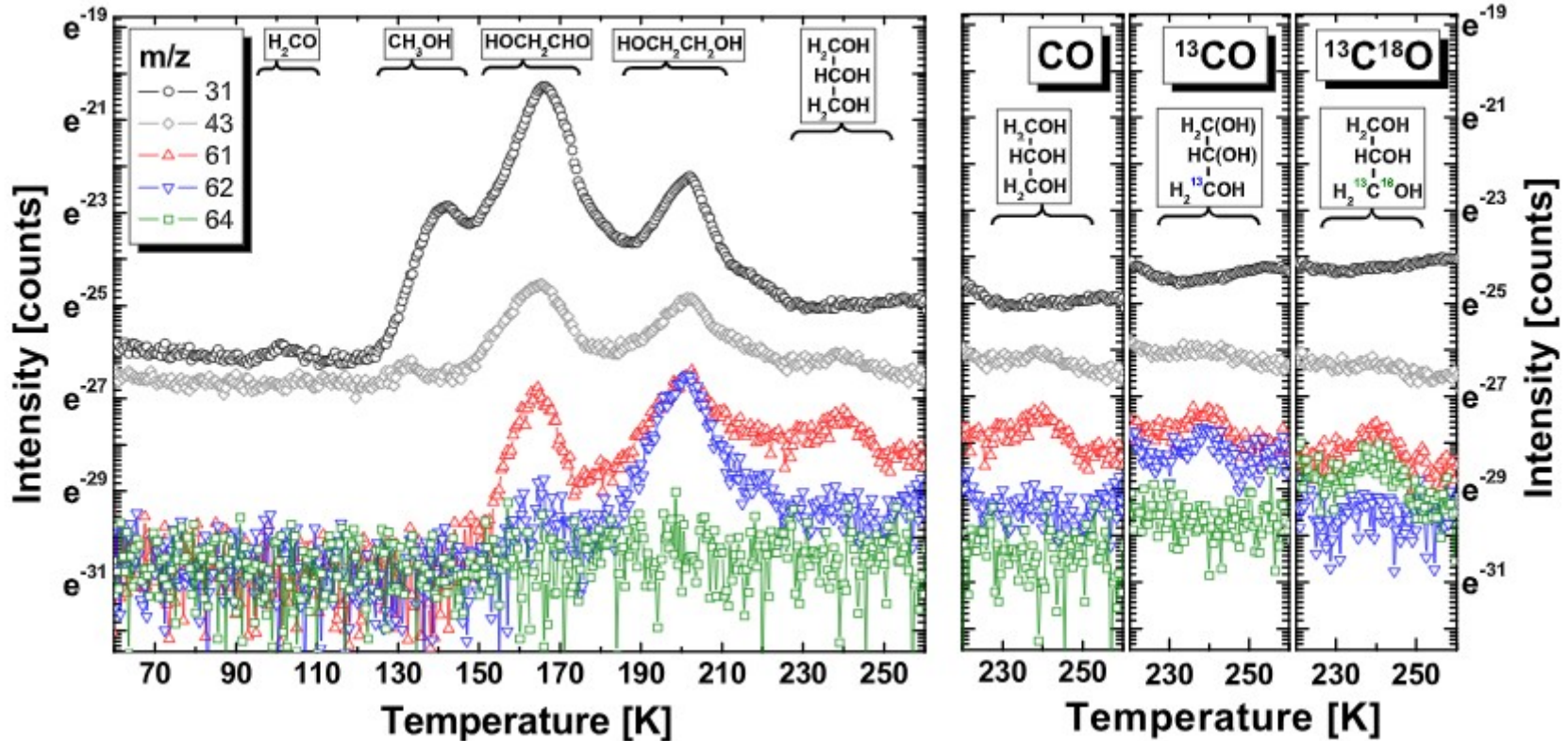
# Grain surface chemistry



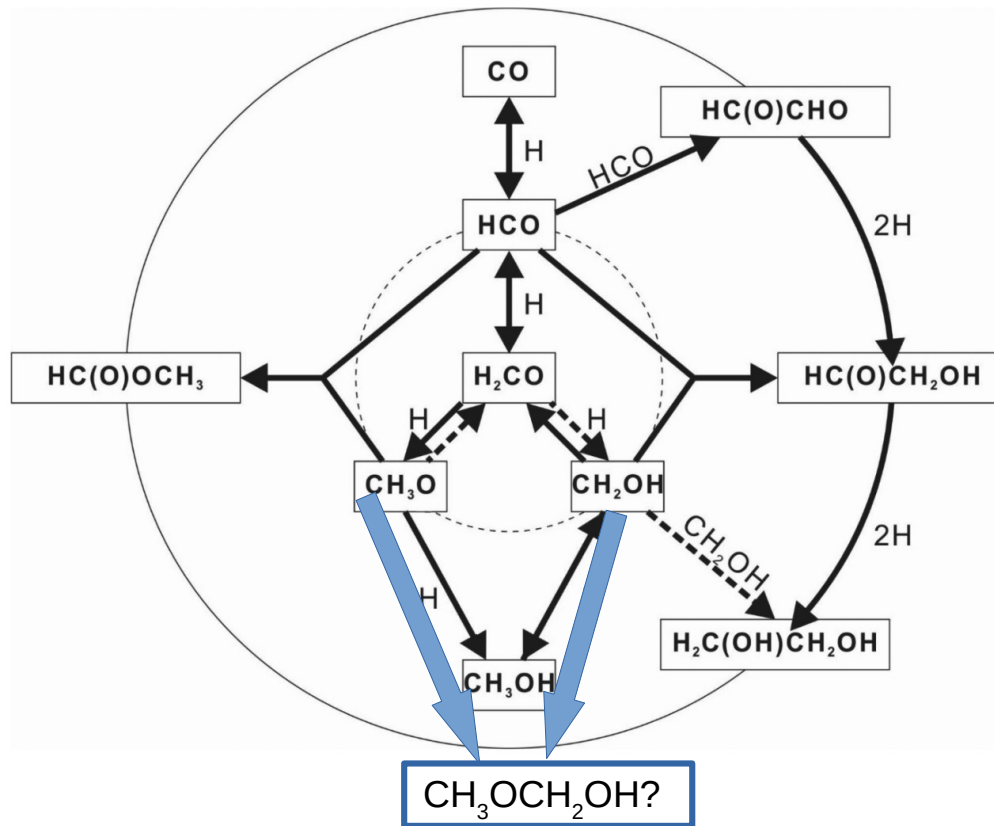
# Chemistry in the CO-rich layer



# Formation of 3-Carbon COM---Glycerol



# Chemistry in the CO-rich layer



# Detection of Methoxymethanol

THE ASTROPHYSICAL JOURNAL LETTERS, 851:L46 (8pp), 2017 December 20

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<https://doi.org/10.3847/2041-8213/aaa0c3>

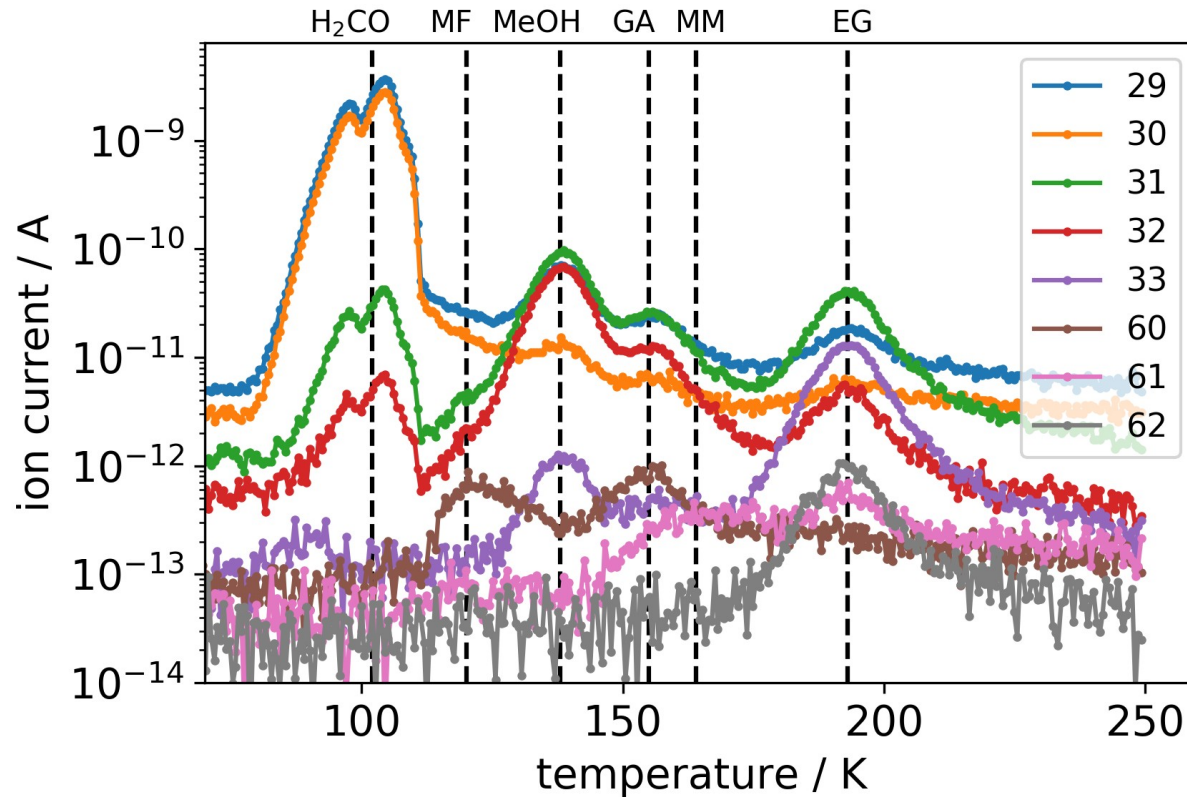


## ALMA Detection of Interstellar Methoxymethanol ( $\text{CH}_3\text{OCH}_2\text{OH}$ )

	$\text{CH}_3\text{OH}:\text{CH}_3\text{OCH}_2\text{OH}$
Observation	34:1
Modeling	$10^7:1$



# Formation of methoxymethanol by $\text{H}+\text{H}_2\text{CO}$



H:H<sub>2</sub>CO=18:1 at 10 K + TPD

He+, in prep

# Prior experimental study of methoxymethanol formation

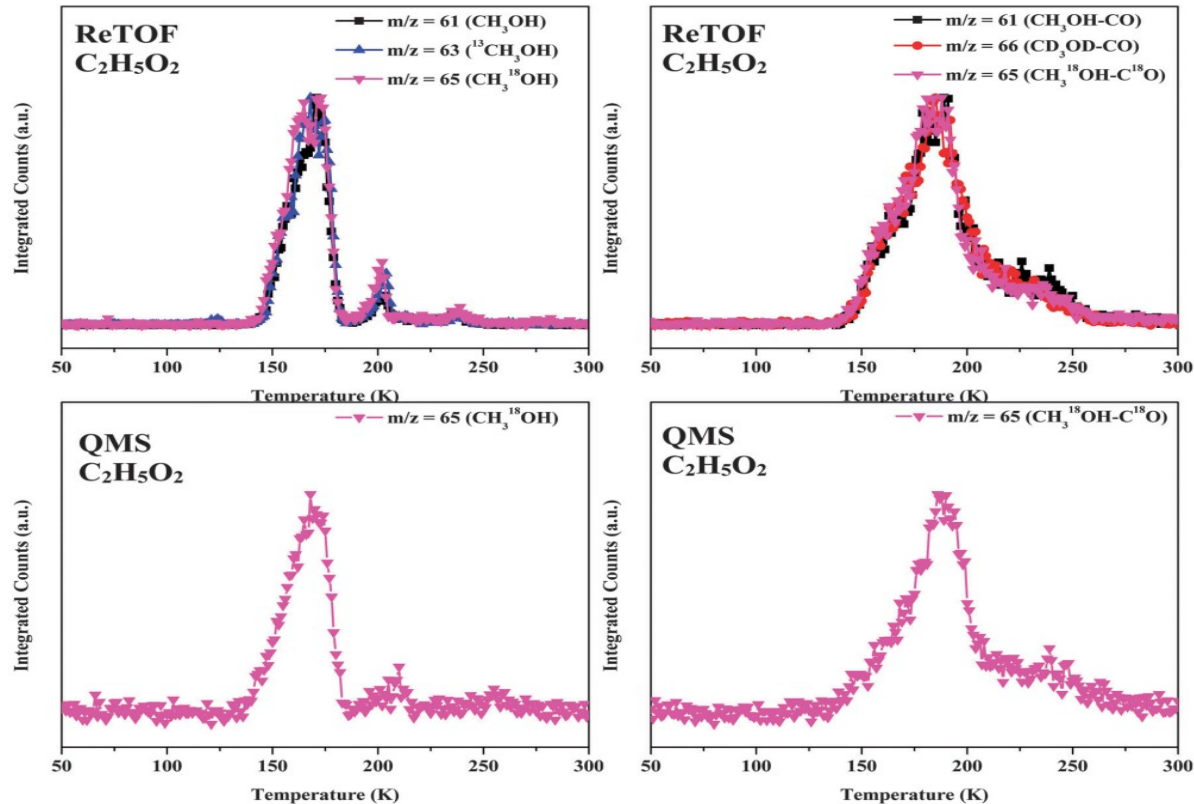
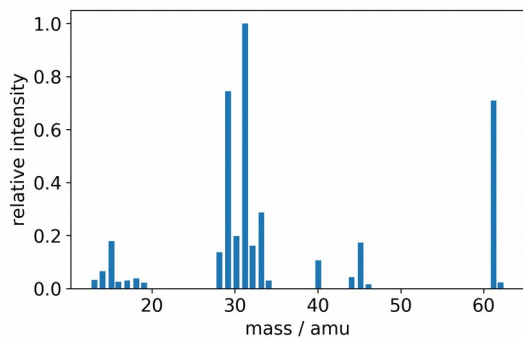
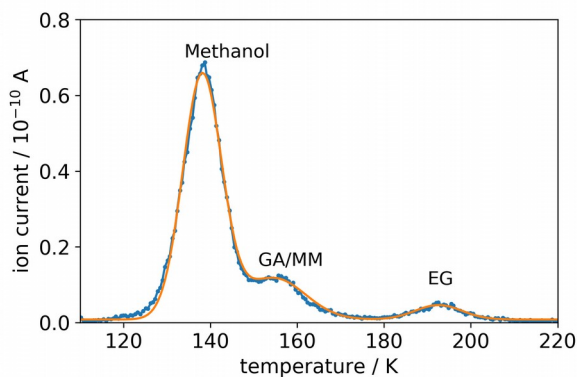
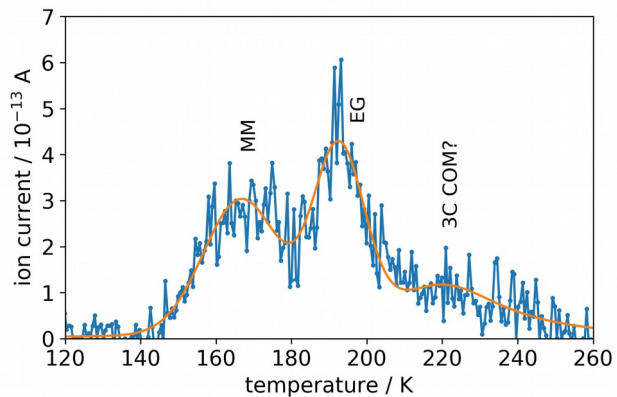
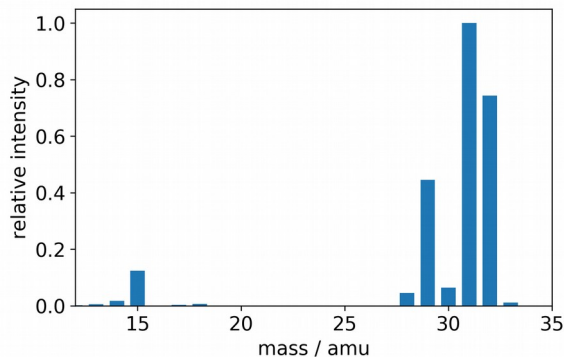


Fig. 8 ReTOF sublimation profiles and QMS traces of  $C_2H_5O_2$  isotomers of irradiated methanol ice systems (left) and methanol-carbon monoxide systems (right). Methoxy methanol ( $CH_3OCH_2OH$ ) is assigned to the observed ion signal.

# Calculate the MeOH:MM ratio



MM:  $m_{61}/\text{total}=0.18$



MeOH:  $m_{32}/\text{total}=0.30$

Experiment  
MeOH:MM = 80

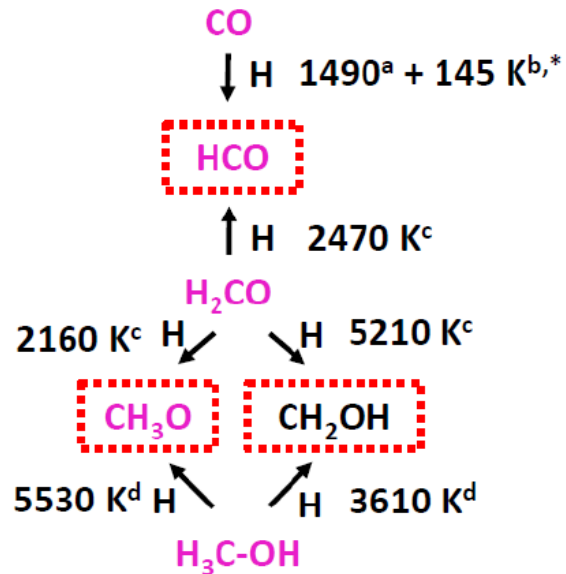
Observation  
MeOH:MM = 34

Modeling  
MeOH:MM =  $10^7$

$$(total\ ion\ counts) \propto (number\ of\ molecules) \times (ionization\ cross-section)$$

# Some remaining questions

Formation of HCO, CH<sub>2</sub>OH,  
and CH<sub>3</sub>O radicals



- What is the branching ratio of H+H<sub>2</sub>CO?
- What is the role of reactive desorption?
- Formation of COMs at 10 K versus during warming up

## Conclusions

- Bottom-up atom addition reactions can build up the chemical complexity in interstellar ices.
- H+CO reaction network in the CO-rich layer is able to explain the formation of methoxymethanol in ISM.

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- Herma Cuppen
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